Tautomeric Structure of Dihydropyrazolo[1,5-a]pyrimidin-7-one in a Solution. A Facile Synthesis of Novel 6-Quinoxalinyldihydropyrazolo[1,5-a]pyrimidin-7-ones by Ring Transformation

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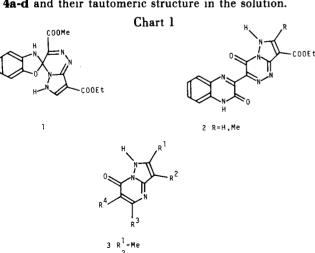
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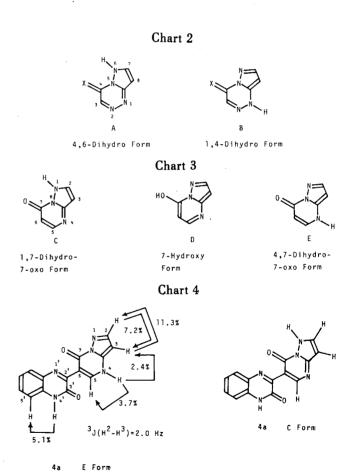
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6-Quinoxalinyldihydropyrazolo[1,5-a]pyrimidin-7-ones **4a-d** were synthesized by the ring transformation of 3-(N,N-dimethylcarbamoyl)furo[2,3-b]quinoxaline hydrochloride **5**. Compounds **4a-d** were found to exist as the 4,7-dihydro-7-oxo form **E** in dimethyl sulfoxide based on the nmr spectral data.

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In a previous paper [2], we reported that the dihydropyrazolo[5,1-c][1,2,4]triazines 1,2 (Chart 1) synthesized by us [2,3,4] predominated as the 4,6-dihydro form A, but not the 1,4-dihydro form B, in the deuteriodimethyl sulfoxide (DMSO-d₆) solution (Chart 2). On the other hand, the dihydropyrazolo[1,5-a]pyrimidin-7-ones 3 (Chart 1) were reported to exist as a mixture of the 1,7-dihydro-7-oxo form C and 7-hydroxy form D (Chart 3) in the solid state [5]. However, there have been few papers concerning the tautomeric structure of dihydropyrazolo[1,5-a]pyrimidin-7ones in a solution. In continuation of the above works, we found that the 6-quinoxalinyldihydropyrazolo[1,5-a]pyrimidin-7-ones 4a-d (Scheme 1) synthesized by the ring transformation of 3-(N,N-dimethylcarbamoyl)furo[2,3-b]quinoxaline hydrochloride 5 [6] were predominant as the 4,7-dihydro-7-oxo form E (Chart 3) in the DMSO-d₆ solution. This paper describes a convenient synthesis of novel 6-quinoxalinyldihydropyrazolo[1,5-a]pyrimidin-7-ones 4a-d and their tautomeric structure in the solution.





The reaction of 3-(N,N-dimethylcarbamoyl)furo[2,3-b]-quinoxaline hydrochloride 5 with the 5-amino-1H-pyrazoles 6a-d in the presence of pyridine gave the 6-quinoxalinyldihydropyrazolo[1,5-a]pyrimidin-7-ones 4a-d, respectively, but not the 6-quinoxalinyldihydropyrazolo[1,5-a]pyrimidin-5-ones 7 (Scheme 1). The formation of 7 was eliminated by the following results. Namely, the reaction of 5 with the 5-amino-1-methylpyrazoles 8 in

Scheme 1

the presence of pyridine did not afford 3-[N-(1-methylpyrazol-5-yl)carbamoyl]furo[2,3-b]quinoxalines 9, but recovered the free base of 5. Moreover, the one dimensional nOe different spectrum of 4a in DMSO-d, supported the structure 4, especially the 4,7-dihydro-7-oxo form 4E (Scheme 1, Chart 3). The radiation at the N₄-H and N₄-H proton signals (both observed at δ 12.50 ppm) showed the 2.4%, 3.7% and 5.1% nOe to the C₃-H, C₅-H and $C_{s'}$ -H proton signals (δ 6.29, 8.32 and 7.35 ppm), respectively, but exhibited no nOe to the C₂-H proton signal (δ 7.96 ppm) (Chart 4), excluding the tautomeric structure of the 1,7-dihydro-7-oxo form 4C (Scheme 1, Chart 3). The radiation at the C2-H and C3-H proton signals showed the 7.2% and 11.3% nOe to the C3-H and C₂-H proton signals, respectively. The coupling constant between the C₂-H and C₃-H protons in the pyrazole ring of 4a was 2.0 Hz, which was a similar value to that of the 1H-pyrazoles 10 (1.9-2.0 Hz) [7] (Chart 5), but not to that of the 1,2-dimethylpyrazolones 11 (3.50-3.60 Hz) [8], furnishing an additional evidence for the tautomeric structure \mathbf{E} , but not \mathbf{C} . The signals due to the \mathbf{C}_2 , \mathbf{C}_{3a} , \mathbf{C}_5 , \mathbf{C}_6 and \mathbf{C}_7 carbons were similar among compounds $\mathbf{4a-c}$ (Table). These data indicated that the 6-quinoxalinyldihydropyrazolo[1,5-a]pyrimidin-7-ones $\mathbf{4a-d}$ existed as the 4,7-dihydro-7-oxo form \mathbf{E} in a solution.

Table

13C-NMR Spectral Data for Compounds 4a-c [a]

Carbon	Compound 4a	Compound 4b	Compound 4c
C ₂	143.47 (dd, J = 184.0, J = 4.5)	145.38 (d, J = 194.3)	143.42 (d, J = 190.0)
C ₃	89.96 (dd, J = 179.0, J = 10.5)	76.31 (d, J = 10.5)	97.57 (d, J = 10.0)
C3.	141.39 (ddd, $J = 175.0$, $J = 10.0$) J = 3.5, J = 8.5)	145.59 (dd, $J = 10.0$, J = 5.0)	142.86 (dd, $J = 9.0$, $J = 4.5$)
C ₅	140.45 [b]	142.33 [b]	141.28 (d, J = 184.0)
C ₆	105.30 (d, J = 2.5)	108.03 (d, J = 2.5)	108.57 (d, J = 2.0)
C ₇	154.86 (d, J = 8.3)	154.16 (d, J = 8.5)	154.00 (d, J = 8.0)
C ₂ ,	153.94 (d, J = 3.5)	152.69 (d, J = 3.5)	152.86 (d, J = 3.5)
C ₃ ,	154.31 (s)	154.16 (s)	154.19 (s)

EXPERIMENTAL

All melting points were determined on an Ishii melting point apparatus and are uncorrected. The ir spectra (potassium bromide) were recorded with a JASCO IRA-1 spectrophotometer. The nmr spectra were measured in deuteriodimethyl sulfoxide with a VXR-300 spectrometer at 300 MHz. Chemical shifts are given in the δ scale. The mass spectra (ms) were determined with a JEOL JMS-01S spectrometer. Elemental analyses were performed on a Perkin-Elmer 240B instrument.

6-(3-0xo-3,4-dihydroquinoxalin-2-yl)-4,7-dihydropyrazolo[1,5-a]-pyrimidin-7-one **4a**, 6-(3-0xo-3,4-dihydroquinoxalin-2-yl)-3-cyano-4,7-dihydropyrazolo[1,5-a]pyrimidin-7-one **4b** and 6-(3-0xo-3,4-dihydroquinoxalin-2-yl)-3-cyano-2-methyl-4,7-dihydropyrazolo-[1,5-a]pyrimidin-7-one **4d**.

General Procedure.

A suspension of 5 (5 g, 18.0 mmoles) and the appropriate pyrazole 6a (2.24 g), 6b (2.92 g) or 6d (3.29 g) (27.0 mmoles, 1.5-fold) in pyridine (5 ml)/1-butanol (200 ml) was refluxed in an oil bath for 2 hours to precipitate yellow needles 4a, brick red needles 4b or orange needles 4d, respectively, which were collected by suction filtration. Trituration with hot ethanol gave analytically pure samples, yield, 4a (3.31 g, 66%), 4b (4.45 g, 81%), 4d (4.03 g, 70%).

Compound 4a.

This compound had mp above 310°; ir: ν cm⁻¹ 3280, 3220, 1675, 1605; ms: m/z 279 (M*); pmr: 12.50 (s, 2H, N₄-H and N₄-H), 8.32 (s, 1H, C₅-H), 7.96 (d, J = 2.0 Hz, 1H, C₂-H), 7.81-7.28 (m, 4H, aromatic), 6.29 (d, J = 2.0 Hz, 1H, C₅-H).

Anal. Calcd. for C₁₄H₉N₅O₂: C, 60.21; H, 3.25; N, 25.08. Found: C, 60.21; H, 3.45; N, 25.26.

Compound 4b.

This compound had mp above 310°; ir: ν cm⁻¹ 3100, 3020, 2230, 1670, 1650; ms: m/z 304 (M⁺); pmr: 12.60 (s, 2H, N₄-H and N₄-H), 8.44 (s, 2H, C₅-H and C₂-H), 7.82-7.30 (m, 4H, aromatic). Anal. Calcd. for C₁₅H₈N₅O₂: C, 59.21; H, 2.65; N, 27.62. Found:

C, 59.01; H, 2.87; N, 27.38.

Compound 4d.

This compound had mp above 310°; ir: v cm-1 3170, 3120,

3060, 2230, 1680, 1670; ms: m/z 318 (M $^+$); pmr: 12.53 (s, 2H, N₄-H and N₄-H), 8.44 (s, 1H, C₅-H), 7.82-7.29 (m, 4H, aromatic), 2.43 (s, 3H, CH₃).

Anal. Calcd. for $C_{16}H_{10}N_6O_2$: C, 60.38; H, 3.17; N, 26.40. Found: C, 60.10; H, 3.16; N, 26.13.

6-(3-Oxo-3,4-dihydroquinoxalin-2-yl)-3-ethoxycarbonyl-4,7-dihydropyrazolo[1,5-a]pyrimidin-7-one 4c.

A solution of **5** (5 g, 18.0 mmoles) and the pyrazole **6c** (4.19 g, 27.0 mmoles) in pyridine (5 ml)/N,N-dimethylformamide (100 ml) was refluxed in an oil bath for 3 hours. Evaporation of the solvent in vacuo afforded yellow crystals **4c**, which were triturated with ethanol and then collected by suction filtration. Recrystallization from N,N-dimethylformamide/ethanol afforded yellow needles (4.54 g, 74%), mp above 310°; ir: ν cm⁻¹ 3260, 1680; ms: m/z 351 (M*); pmr: 12.55 (s, 2H, N₄-H and N₄-H), 8.30 (s, 1H, C₅-H), 8.27 (s, 1H, C₂-H), 7.79-7.33 (m, 4H, aromatic), 4.34 (q, J = 7 Hz, 2H, CH₂), 1.34 (t, J = 7 Hz, 3H, CH₃).

Anal. Calcd. for C₁₇H₁₃N₅O₄: C, 58.12; H, 3.73; N, 19.94. Found: C, 58.07; H, 3.75; N, 19.83.

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REFERENCES AND NOTES

- [1] Present address: Department of Chemistry, Teacher's College, Hyosung Women's University, Gyongsan 713-900, Korea.
- [2] Y. Kurasawa, Y. Kamigaki, H. S. Kim, K. Yonekura, A. Takada and Y. Okamoto, J. Heterocyclic Chem., in press.
- [3] Y. Kurasawa, M. Okiyama, Y. Kamigaki, M. Kanoh, Y. Okamoto and A. Takada, J. Heterocyclic Chem., 24, 1805 (1987); Y. Kurasawa, M. Okiyama, Y. Kamigaki, M. Kanoh, A. Takada and Y. Okamoto, ibid., 25, 1015 (1988).
- [4] Y. Kurasawa, A. Satoh, S. Ninomiya, H. Arai, K. Arai, Y. Okamoto and A. Takada, J. Heterocyclic Chem., 24, 1229 (1987).
- [5] I. Hori, K. Saito and H. Midorikawa, Bull. Chem. Soc. Japan, 43, 849 (1970).
- [6] Y. Kurasawa and A. Takada, Heterocycles, 14, 281 (1980); idem, Chem. Pharm. Bull., 29, 2871 (1981).
- [7] H. Dorn and D. Arndt, Ann. Chem., 750, 39 (1971); J. Elguero, R. Jacquier and H. C. N. Tien Duc, Bull. Soc. Chim. France, 3727 (1966).
 - [8] M. Begtrup, Acta Chem. Scand., 24, 1819 (1970).